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100 Viva-voce questions on volumetric analysis

100 Viva-voce questions on volumetric analysis (a) General 1 What is the principle of volumetric analysis? 2 What is titration? 3 Why burette and pipette must be ...

VERY SHORT ANSWER QUESTIONS 2. - Sakshi Education

Induced catalyst: When one reaction influences the rate of other reaction, which does not occur under ordinary conditions, the phenomenon is known as induced catalysis SHORT ANSWER QUESTIONS 1 Explain the types of adsorptions Adsorption: The phenomenon in which the molecules of a ...

Chemistry - Chhatrapati Shahu Ji Maharaj University

Chemical kinetics and its scope, rate of a reaction, factors influencing the rate of a reaction - concentration, temperature, pressure, solvent, light catalyst, concentration dependence of rates, mathematical characteristics of simple chemical reactions - ...

ENGINEERING CHEMISTRY CH-103-F LAB MANUAL I, II ...

world college of technology and management, gurgaon (haryana) engineering chemistry ch-103-f lab manual i, ii semester b tech

IV SEMMESTER

This reaction follows pseudo first order kinetics PROCEDURE: 100 ml of 0.5 N HCl is taken in a clean dry conical flask 5 ml of ester is pipetted out into the conical flask and the mixture is immediately withdrawn into another dry conical flask A stop watch is started simultaneously The reaction is then arrested by the addition of

Unit 6 Subjects INTRODUCTION TO VOLUMETRIC ANALYSIS

Titration Reaction Not any chemical reaction can be used in the volumetric analysis There are some conditions to be met in order for a chemical reaction to be used as a basis for a titration : 1- The reaction must proceed according to a definite balanced chemical equation ie no side reactions In other word it must be stoichiometric

EXPERIMENT - 1 Estimation of Hardness of Water by EDTA ...

EXPERIMENT - 1 Estimation of Hardness of Water by EDTA Method INTRODUCTION Water hardness is the traditional measure of the capacity of water to precipitate soap Hard water requiring a considerable amount of soap to produce leather Scaling of hot water pipes, boilers and other house Viva Questions Title: untitled Created Date:

COMMON DISTINCTION TESTS IN ORGANIC CHEMISTRY

Grade 12 COMMON DISTINCTION TESTS IN ORGANIC CHEMISTRY I R - Cl vs R - Br vs R - I (R alkyl or aryl) SNo Test R - Cl R - Br R - I a) Dil AgNO₃ b) OH NH₄ test II Ethylidene chloride (Geminal) vs Ethylene Dichloride (Vicinal) SNo Test

Precipitation Titration: Determination of Chloride by the ...

Precipitation Titration: Determination of Chloride by the Mohr Method by Dr Deniz Korkmaz Introduction Titration is a process by which the concentration of an unknown substance in solution is determined by adding measured amounts of a standard solution that reacts with the unknown

PREPARATION OF ACETYLSALICYLIC ACID (ASPIRIN)

4 Post-lab Questions a) What was used to increase the rate of the reaction? b) What would happen to your percent yield if the aspirin you prepared was not dried completely? c) A student's theoretical yield was 120g of aspirin, but he only obtained 75g What was ...

CHEMISTRY PROJECT On Determination Of The Contents Of ...

CHEMISTRY PROJECT On Determination Of The Contents Of Cold Drinks PROJECT PREPARED BY: XII various questions regarding their purity News flashed that they contain harmful pesticide, which CHEMICAL REACTION INVOLVED $\text{NaHPO}_4 + 12(\text{NH}_4)_2\text{MoO}_4 + 21\text{HNO}_3 + 3\text{H}^+ \rightarrow (\text{NH}_4)_3\text{PO}_4 \cdot 12\text{MoO}_3$

General Organic Chemistry Questions

Organic Chemistry Questions The Covalent Bond 1 The hybridization of the central carbon in $\text{CH}_3\text{C}\equiv\text{N}$ and the bond angle CCN are a) sp², 180° b) sp, 180° c) sp², 120° d) sp³, 109° 2 Which of the following statements about an sp hybridized carbon is FALSE?

B.Sc. - FIRST YEAR

BSc - FIRST YEAR CHEMISTRY There shall be three written papers and a practical examination as follows: Max Marks Paper - I Inorganic Chemistry 50 Paper - II Organic Chemistry 50 Paper - III Physical Chemistry 50 TOTAL 150 PRACTICAL 50 GRAND TOTAL 200 Candidate will be required to pass in Theory and Practical Separately

2.37H TITRATIONS WITH SODIUM THIOSULFATE

Iodine, the reaction product, is ordinarily titrated with a standard sodium thiosulfate solution, with starch serving as the indicator: I₂ + 2S₂O₃²⁻ → 2I⁻ + S₄O₆²⁻
A discussion of thiosulfate methods is found in Section 20B-2 37H-1 Preparation of 0.1 M Sodium Thiosulfate PROCEDURE Boil about 1 L of distilled water for 10 to 15 min Allow the water to

Determination of an Equilibrium Constant for the Iron (III) ...

4-2 Determination of an Equilibrium Constant for the Iron(III) Thiocyanate Reaction Since the product, FeSCN²⁺, has a deep red color, its concentration can be determined using spectrophotometric techniques-that is, based on how much light it is absorbing

TOPIC 11. OXIDATION AND REDUCTION REACTIONS.

TOPIC 11 OXIDATION AND REDUCTION REACTIONS The reverse reaction, conversion of compounds such as oxides of metals to the elemental metal were called "reduction" reactions, for example, the reduction of At the beginning of the course, the concepts of compounds and of chemical ...